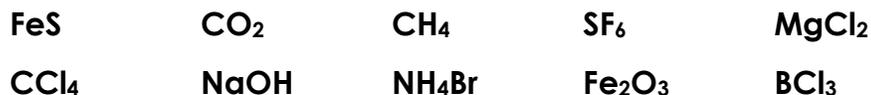


**Chemistry**  
**Molecular Compounds Names & Formulas (POINTS)**

Name \_\_\_\_\_

1. Circle the formulas for the molecular compounds in the list below:



2. Fill in the chart below with the correct Greek prefixes.

1	2	3	4	5	6	7	8	9	10

*Instructions: Write the scientific name of the compound on the line next to the formula.*

3. SiO<sub>2</sub> \_\_\_\_\_  
4. H<sub>2</sub>O \_\_\_\_\_  
5. N<sub>2</sub>O<sub>5</sub> \_\_\_\_\_  
6. N<sub>2</sub>O<sub>3</sub> \_\_\_\_\_  
7. CS<sub>2</sub> \_\_\_\_\_  
8. F<sub>2</sub>Cl<sub>5</sub> \_\_\_\_\_  
9. SO \_\_\_\_\_  
10. SO<sub>2</sub> \_\_\_\_\_  
11. PI<sub>5</sub> \_\_\_\_\_  
12. N<sub>4</sub>I<sub>10</sub> \_\_\_\_\_

*Instructions: Write the formula of the compound on the line next to the name.*

13. Carbon monoxide \_\_\_\_\_  
14. Carbon tetrachloride \_\_\_\_\_  
15. Dinitrogen Trioxide \_\_\_\_\_  
16. sulfur hexafluoride \_\_\_\_\_  
17. silicon dioxide \_\_\_\_\_  
18. phosphorus tribromide \_\_\_\_\_  
19. Nitrogen trihydride \_\_\_\_\_  
20. Tetrasulfur dinitride \_\_\_\_\_  
21. Dichlorine heptafluoride \_\_\_\_\_  
22. Oxygen pentabromide \_\_\_\_\_