

# More Practice Polarity WS

Instructions: Fill out the table for each of the compounds.

Formula	Lewis Structure	Electronegativity Difference	Molecular Shape	Different outside atoms OR lone pairs on central atom? (Y/N)	Polar? (Y/N)
XeF <sub>4</sub>					
SF <sub>4</sub>					
HF					
O <sub>2</sub>					

Formula	Lewis Structure	Electronegativity Difference	Molecular Shape	Different outside atoms OR lone pairs on central atom? (Y/N)	Polar? (Y/N)
<b>BrF<sub>3</sub></b>					
<b>CO<sub>3</sub><sup>-2</sup></b>					
<b>I<sub>3</sub></b>					
<b>AsF<sub>5</sub></b>					