Naming &	Writing	Acids &	Bases	(Points)
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Instructions: Write the letter "A	" for acid or "B" for base	e on the line next to the formula.
1. HCl	5. Fe(OH)₃	9. H ₃ PO ₃
2. LiOH	6. HClO ₄	10. AI(OH) ₃
3. HNO ₂	7. HBrO ₃	
4. H ₂ S	8. Sr(OH) ₂	
Instructions: Write the scientific	name of the compour	nd on the line next to the formula.
1. HF		<u> </u>
2. HClO ₄		_
3. HI		<u> </u>
4. H ₂ CO ₃		_
5. CuOH		<u> </u>
6. HMnO ₄		<u> </u>
7. HNO ₂		<u> </u>
8. LiOH		<u> </u>
9. Pb(OH) ₂		_
10.NH ₄ OH		_
Instructions: Write the formula	of the compound on th	ne line next to the name.
1. Potassium hydroxide		
2. Bromic Acid		
3. Hydroselenic Acid		
4. Phosphorous Acid		
5. Hypochlorous Acid		
6. Sulfuric Acid		
7. Hydrofluoric Acid		
8. Calcium hydroxide		
9. Lead (IV) hydroxide		
10. Nitrous acid		