Instructions: Write the TWO half reactions from the following equations. Next, determine which equation was oxidation and which was reduction.

1) $Cu + NO_3^{1-} \rightarrow Cu^{2+} + NO_2$

2) $I_2(s) + OCI^{-1} \rightarrow IO_3^{-1} + CI^{-1}$

3) $TeO_3^{2-} + N_2O_4 \rightarrow Te + NO_3^{-1}$

4) $H_2O_2 + Cr_2O_7^2 \rightarrow O_2 + Cr^{3+}$